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MULTIFUNCTIONAL NEUTRAL PHOSPHENIUM ION COMPLEXES

HEINRICH LANG

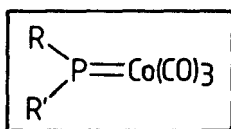
Anorganisch-Chemisches Institut der Universität, Im
 Neuenheimer Feld 270, D 6900 Heidelberg 1, Germany

Abstract

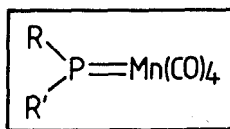
The synthesis, structural features and reactivity of the neutral phosphonium ion complexes $(R)(R')P=ML_n$ ($ML_n = Co(CO)_3$, $Mn(CO)_4$, $MoCp(CO)_2$; $R = 2,4,6\text{-}^tBu_3C_6H_2O$; $R' = C\equiv CPh$, $CH=CHPh$, $CH_2C\equiv CH$, $CH=C=CH_2$) will be discussed.

Synthesis

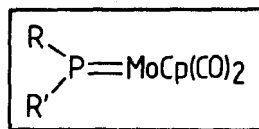
Complexes of the type $(R)(R')P=ML_n$ (ML_n , R , R' = see above), 1 - 3, can be prepared either via reductive dehalogenation of $(R)(R')PCl$ with $Co_2(CO)_8$ ¹ or via NaCl metathesis reaction of $(R)(R')PCl$ with $NaML_{n+1}$ ($ML_{n+1} = Co(CO)_4$ ¹, $Mn(CO)_5$ ², $MoCp(CO)_3$ ³). Decarbonylation of $(R)(R')P-ML_{n+1}$ affords then 1 - 3. 1 and 2 are the first examples of this type of compounds with a phosphorus-cobalt (1) or a phosphorus-manganese (2) π -bond.



1



2



3

Structure, Spectra

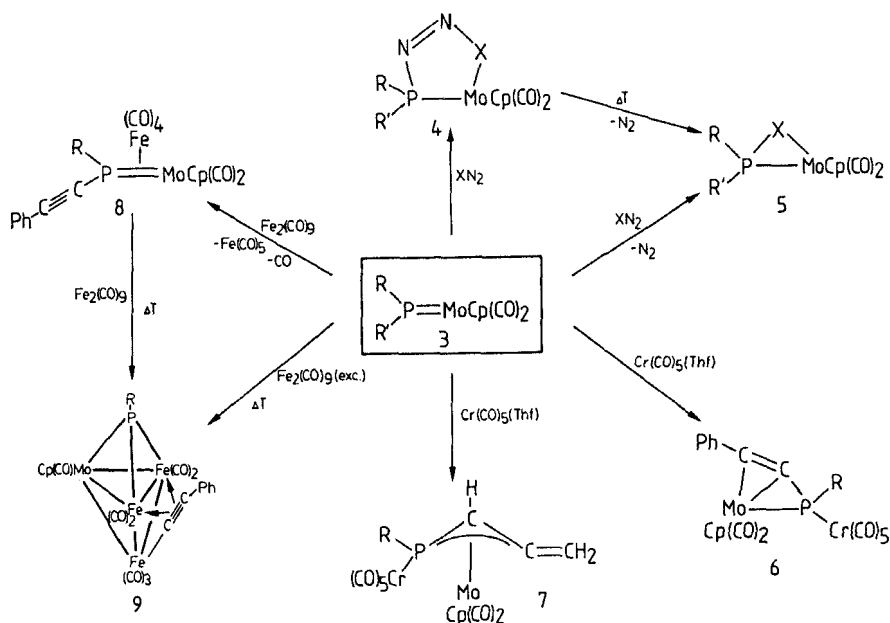
The trigonal planar coordination with short P—M bond lengths is common to all $(R)(R')P=ML_n$ compounds 1 - 3¹⁻⁸. Such bond shortening can be taken to imply $P=M$ multiple bonding in 1 - 3.

1 - 3 show bright colours due to their $P=M$ multiple bond. The observed down field shift in the ^{31}P -NMR spectra of 1 - 3 is typical for complexes with a sp^2 -hybridised phosphorus atom⁴.

Reactivity

Some typical reactions of 1 - 3, with CH_2N_2 , PhN_3 and organometallic complex fragments, isolobal to carbene, are shown in scheme 1⁵.

1,3-dipolar cycloaddition of CH_2N_2 or PhN_3 with the formal PMo double bond yields the five-membered cycles of 4. On warming up 4 eliminates N_2 yielding 5 (scheme 1).

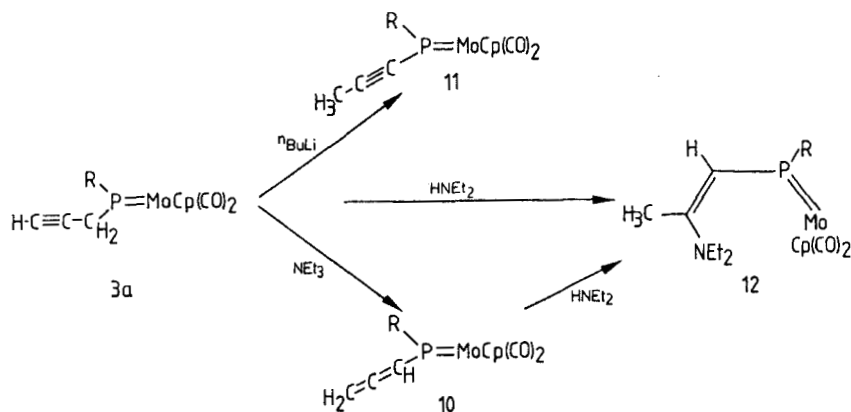


Scheme 1: Reaction of 3 with XN_2 ($\text{X} = \text{CH}_2$, NPh), $\text{Fe}_2(\text{CO})_9$ and $(\text{CO})_5\text{Cr}(\text{Thf})$.

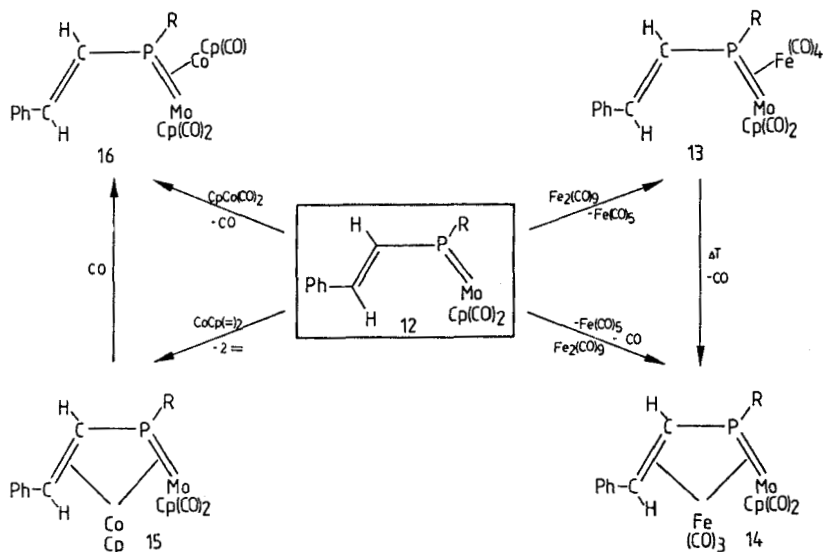
An easy access to 1-phosphaallyl complexes (7) is given by the propargyl-allenyl isomerization in $(\text{R})(\text{HC}\equiv\text{CCH}_2)\text{P}=\text{MoCp}(\text{CO})_2$, 3a, induced by $\text{Cr}(\text{CO})_5(\text{Thf})$ in Thf, whereas the reaction of the phenylethynyl substituted derivative 3b yields 6 (scheme 1)⁵. However, with $\text{Fe}_2(\text{CO})_9$ the addition of $\text{Fe}(\text{CO})_4$ to the formal PMo double bond is observed (8; scheme 1)⁵. If an excess of $\text{Fe}_2(\text{CO})_9$ is used, the phosphorus-alkynyl-carbon- σ -bond in 3b is broken and cluster 9 is formed (scheme 1)⁵.

Another possibility for initiating the 2-propynyl isomerization in 3a is given by the reaction with amines or *n*-BuLi: in the presence of NEt_3 the allenyl substituted compound 10 is obtained, whereas with *n*-BuLi the corresponding 1-propynyl substituted derivative 11 can be isolated (scheme 2)⁶.

The reaction of 3a and 10 with HNEt_2 leads via a hydroamination type reaction stereoselectively to the heterobutadiene 12 (scheme 2)⁶. Compounds similar to 12 can also be synthesized by the reaction of $(R)(R')\text{PCl}$ ($R' = \text{CH=CHPh}$, CH=CMe_2 , ...) with $\text{Na}[\text{Cp}(\text{CO})_3\text{Mo}]^{2b}$.



Scheme 2: Propargyl isomerization of 3a.

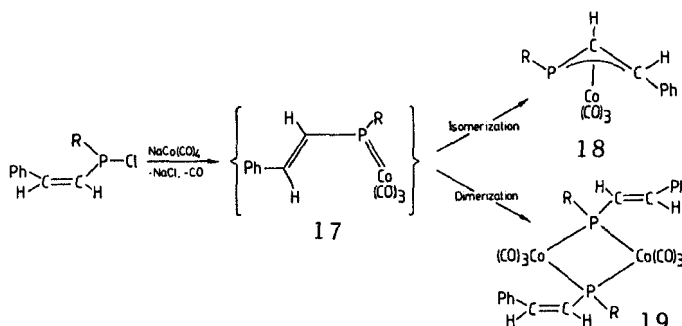


Scheme 3: Reaction of 12 with $\text{Fe}_2(\text{CO})_9$ and $\text{CoCp}(\text{H}_2\text{C=CH}_2)_2$.

To prove the butadiene analogy of 12, it was reacted with $\text{Fe}_2(\text{CO})_9$ and $\text{CpCo}(\text{H}_2\text{C}=\text{CH}_2)_2$ as shown in scheme 3⁷.

12 yields with $\text{Fe}_2(\text{CO})_9$ 13; 13 decarbonylates at 60 °C to 14⁷. In this compound the heterobutadiene unit is η^4 -side-on coordinated to $\text{Fe}(\text{CO})_3$. In a similar way 12 reacts with $\text{CpCo}(\text{H}_2\text{C}=\text{CH}_2)_2$ to 15 (scheme 3)⁷. In the presence of carbon monoxide 15 forms 16, a compound, in which only the PMo multiple bond is η^2 -side-on bonded to a $\text{CpCo}(\text{CO})$ moiety⁷.

A possible way to synthesize the 1-cobalta-2-phospha-1,3-diene 17 is given in scheme 4⁸.



Scheme 4: Isomerization, dimerization of 17.

(R)(PhHC=CH)P(Cl)R reacts with $\text{NaCo}(\text{CO})_4$ to yield 18 and 19⁸. For the formation of 18 and 19 an intermediate like 17 is probably responsible. From 17 the formation of 18 and 19 may easily be explained: isomerization yields 18 and dimerization of the PCo multiple bond 19 (scheme 4)⁸.

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